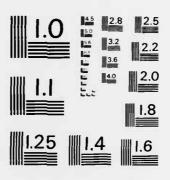
THEORETICAL ASPECTS OF LASER-INDUCED PERIODIC SURFACE STRUCTURE FORMATION(U) ROCHESTER UNIV NY DEPT OF CHEMISTRY M HUTCHINSON ET AL. NOV 83 UROCHESTER/DC/83-TR-43 N00014-80-C-0472 F/G 7/4 AD-A134 875 1// UNCLASSIFIED NL END 12-83 DTIC



MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS - 1963 - A

SECURITY CLASSIFICATION OF THIS PAGE (When Date Entered)



RE..D INSTRUCTIONS BEFORE COMPLETING FORM REPORT DOCUMENTATION PAGE 1. REPORT NUMBER 2. GOVT ACCESSION NO. 3. RECIPIENT'S CATALOG NUMBER UROCHESTER/DC/83/TR-43 4. TITLE (and Subtitle) 5. TYPE OF REPORT & PERIOD COVERED Theoretical Aspects of Laser-Induced Periodic Surface Structure Formation 6. PERFORMING ORG. REPORT NUMBER 7. AUTHOR(e) 8. CONTRACT OR GRANT NUMBER(#) Michael Hutchinson, Ki-Tung Lee, William C. Murphy, A. C. Beri and N00014-80-C-0472 Thomas F. George PERFORMING ORGANIZATION NAME AND ADDRESS 10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS Department of Chemistry University of Rochester NR 056-749 Rochester, New York 14627 11. CONTROLLING OFFICE NAME AND ADDRESS 12. REPORT DATE Office of Naval Research November 1983 Chemistry Program Code 472 13. NUMBER OF PAGES Arlington, Virginia 22217

16. DISTRIBUTION STATEMENT (of this Report)

This document has been approved for public release and sale; its distribution is unlimited.

17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, If different from Report)

14. MONITORING AGENCY NAME & ADDRESS(II dillerent from Controlling Office)

15. SECURITY CLASS. (of this report)

15a. DECLASSIFICATION/DOWNGRADING

Unclassified

16. SUPPLEMENTARY NOTES

Prepared for publication in Laser-Controlled Chemical Processing of Surfaces, ed. by A. W. Johnson and D. J. Ehrlich (Elsevier, New York)

19. KEY WORDS (Continue on reverse elde II necessary and Identily by block number)
PERIODIC SURFACE STRUCTURE FORMATION SMOOTH S LASER-INDUCED THEORETICAL STUDY MAXWELL'S EQUATIONS

SMOOTH SURFACE GRATINGS ROUGH SURFACE GRATINGS MULTILAYERED GRATINGS SQUARE-WELL SILVER GRATING

20. ABSTRACT (Continue on reverse side if necessary and identify by block number] aser-Induced periodic pattern formation has been observed on a variety of substances. In particular, low-power lasers have been used to deposit a pattern on a metal surface. For a relatively smooth surface grating, this pattern can be explained in terms of a perturbative solution of Maxwell's equations. However, as the surface grating is enhanced by this initial deposition, the perturbation solution breaks down. An alternate non-perturbative solution of Maxwell's equations for such rough surfaces is considered here. Moreover, other possible mechanisms that may assist pattern formation are discussed, such as field-enhanced evaporation and surface migration.

DD , FORM , 1473

088 classified

SECURITY CLASSIFICATION OF THIS PAGE (When Dete Entered)

OFFICE OF NAVAL RESEARCH
Contract NO0014-80-C-0472
Task No. NR 056-749
TECHNICAL REPORT No. 43

Theoretical Aspects of Laser-Induced Periodic Surface Structure Formation

by

Michael Hutchinson, Ki-Tung Lee, William C. Murphy, A. C. Beri and Thomas F. George

Prepared for Publication

in

Laser-Controlled Chemical Processing of Surfaces, ed. by A. W. Johnson and D. J. Ehrlich (Elsevier, New York)

Department of Chemistry University of Rochester Rochester, New York 14627

November 1983

Reproduction in whole or in part is permitted for any purpose of the United States Government.

This document has been approved for public release and sale; its distribution is unlimited.

FART 88 1: 88

To appear in <u>Laser-Controlled Chemical Processing of Surfaces</u>, ed. by A. W. Johnson and D. J. Ehrlich (Elsevier, New York)

THEORETICAL ASPECTS OF LASER-INDUCED PERIODIC SURFACE STRUCTURE FORMATION

MICHAEL HUTCHINSON, KI-TUNG LEE, WILLIAM C. MURPHY, A. C. BERI AND THOMAS F. GEORGE Department of Chemistry, University of Rochester Rochester, New York 14627

ABSTRACT

Laser-induced periodic pattern formation has been observed on a variety of substances. In particular, low-power lasers have been used to deposit a pattern on a metal surface. For a relatively smooth surface grating, this pattern can be explained in terms of a perturbative solution of Maxwell's equations. However, as the surface grating is enhanced by this initial deposition, the perturbation solution breaks down. An alternate non-perturbative solution of Maxwell's equations for such rough surfaces is considered here. Moreover, other possible mechanisms that may assist pattern formation are discussed, such as field-enhanced evaporation and surface migration.

INTRODUCTION 120 05 1 2 7 5 2 2 2 2 (12.1 x 19.1 cm).

Over the past several years, a number of researchers [1-17] have observed periodic patterns developing on the surfaces of solids that were exposed to laser radiation. These patterns have been seen on various metals, insulators, and both doped and pure semiconductors. Most of these experiments were carried out with Nd:YAG or ND:glass lasers tuned to wavelengths between 1.0 μ m and 10.0 μ m. To obtain these patterns, laser power densities of between 10 MW/cm² and 1 GW/cm² were employed. Above this range, surface melting would occur, while below this level, no discernible surface pattern was observed.

This laser-induced periodic surface structure has several characteristics. First, the surface pattern consists of a number of parallel grooves running perpendicular to the E field of the incident laser. Second, Oron and Sorensen [8] demonstrated that the underlying surface lattice has no effect on the shape of the laser-induced surface structure. On the other hand, Isenor [6] has shown that the existence of scratches on the surface can have a significant effect on the final surface structure. In fact, if the surface is randomly scratched, no periodic surface structure will form. Both Bruck and Ehrlich [12] and van Driel and his colleagues [13] have also observed the development of a weaker secondary pattern parallel to the incident E field. With this secondary structure, the surface takes on a scalloped appearance. Finally, these surface patterns produce large drops in the reflectivity of several metals [3-4].

Most of the work so far has considered laser-induced periodic structures on surfaces that already had a certain composition. Needless to say, this necessitates large laser power densities (greater than 10 MW/cm²) in order to provide sufficient energy for the surface atoms to rearrange. However, Brueck and Ehrlich [12] showed that only a modest amount of laser power (less-than 10 W/cm²) is needed to establish the periodic surface structure when the constituent atoms are deposited from the gas phase. In particular, they showed that Cd, Zn and Al in organometallic compounds could be photochemically deposited on Si or SiO₂ substrates in ordered arrays.

Accession P.

MTIS GRAFIDTIG TAB
U amouncedJustification

Distribution/
Availability Codes

Aveil and/or

To understand these observations and resolve any difficulties, one must consider how this laser-induced periodic surface structure is produced. There have been three attempts to explain these effects. First, since many metals have a plasma frequency around the frequency of the incident laser radiation, surface plasmons will be excited [11-12]. These plasmons will couple with the electromagnetic field to form a surface polariton. Since this polariton produces maxima and minima in the É field of the surface, the surface atoms will rearrange themselves to minimize their energy in this É field. Second, Maracas et al [7] have observed that the periodic surface structure can be considered a standing wave. This wave would have a velocity very close to that of a longitudinal acoustic phonon of the substrate. the surface pattern may well be a phonon excited by the laser that was "frozen" in place by the cooling of the lattice. Finally, van Driel and his associates [15] have proposed an extensive theory based on the small inhomogeneities that exist in the surface layer. This initial surface roughness will interact with the incident laser beam to produce a dipole moment in the surface layer (a "radiation remnant"). The field generated by this dipole layer can interfere with the refracted beam in the substrate below the surface. This interference will lead to inhomogeneous energy absorption and thus the redistribution of the surface atoms.

In the following section, the theory of the surface enhanced E field produced by relatively smooth surfaces and the resultant pattern formation will be viewed. The inapplicability of this approach to rough gratings will be demonstrated. To overcome this limitation, a non-perturbative solution of Maxwell's equations will be discussed. Finally, our results will be presented along with suggested improvements.

THEORY OF SMOOTH GRATINGS

Incident radiation converted to surface plasmon excitations via surface roughness has been investigated extensively in the past decade [18]. The classical method of Rayleigh [19] has often been used to solve the problem of weak scattering [20]. Brueck and Ehrlich [12] also adapted this formalism to describe the phenomena of laser-induced pattern formation. Here, we will briefly outline the theory of Brueck and Ehrlich and discuss its validation and limitation.

Let a light wave of frequency w represented by an electric field

$$\dot{E}_{inc}(\dot{r},t) = E_i e^{ik_0 z} e^{-i\omega t}$$
 (1)

be incident on the metal surface at a normal angle, on the xz-plane. If we express the surface roughness in terms of a Fourier expansion,

$$z = \xi(x,y) = \sum_{g} u_g e^{ig \cdot r} \Pi$$
 , $g = (g_x, g_y)$, $r_H = (x,y)$, (2)

and assume that each amplitude ug is small compared to the wavelength of the incident light, then taking the dielectric function as a step function,

$$\varepsilon(\omega, r) = \varepsilon(\omega)\Theta(z - \xi) + \Theta(\xi - z), \tag{3}$$

we can write down the solutions of the Maxwell's equations separately for the two media. In particular, for $g^2 > \omega^2/c^2$ and $z > \xi$, we have

$$\vec{E} = \vec{E}_1 e^{-ik_0 z} + \vec{E}_r e^{ik_0 z} + \vec{E}_{sp} e^{i\vec{g} \cdot \vec{r}} || e^{-kg^z},$$
 (4)

where $k_g^2 = g^2 - \omega^2/c^2$. By imposing the boundary conditions, i.e., the local

tangential components of the electric and magnetic fields have to be continuous across the actual surface $z = \xi(x,y)$, we can express the reflected and surface plasmon fields, \tilde{E}_r and \tilde{E}_s respectively, in terms of \tilde{E}_s by writing

$$e^{ik_0\xi} \approx (1 + ik_0\xi) \tag{5}$$

and assuming $g\xi << 1$ in the continuity equations. If we consider only the p-wave scattering, i.e., the light polarized in the x-direction, the expression of Brueck and Ehrlich for $\dot{E}_{\rm p}$ can then be easily obtained. If we assume a linear relationship between the film growth rate and the local intensity,

$$\frac{dT}{dt} \propto I(x) = |\vec{E} \cdot \vec{E}|, \qquad (6)$$

then the suggested equation of the grating depth up is obtained by assuming a spatial profile of $T(x) = T_0 + u \cos(gx)$, and neglecting the $|\vec{E}_{sp}|^2$ term.

The foregoing theory was developed under the assumption that the surface roughness is small, g5 << 1. However, as more metal is deposited, the roughness increases. . In the perturbative theory of Brueck and Ehrlich, E is proportional to the roughness u. Therefore, the first consequence of sp increased roughness is that $|\hat{E}_{s}|^2$ can no longer be neglected. This point may already be reached when the grating is only a few layers deep. Moreover, the damping of E can no longer be measured from the mean surface. An increase in roughness can lead to sufficient enhancement of E that localized evaporation of metal atoms could play an important role. Transfer of large amounts of energy to the metal-carbon bond, such as in the deposition of Cd on Si due to the dissociation of gaseous Cd(CH₃), may also induce sufficient translational motion along the surface so that adsorption does not occur at the dissociation site. Such dissociation under several possibly physisorbed organometallic layers could lead to trapping of the organic radical in the vicinity of the surface. This "cage" effect could lead to the reformation of the organic metal bond. Finally, the effect of surface roughness on the dielectric function might be important. Such increases of surface area could substantially alter the magnitude of this function, which in turn could enhance the rate of pattern formation.

GENERALIZATION TO ROUGH GRATINGS

The foregoing considerations lead us to conclude that an explanation of deep pattern formation requires a non-perturbative treatment of plasmon formation, and at the present time a numerical solution of Maxwell's equations seems inevitable. Recently [21] there has appeared a solution of the problem of a square-well metallic grating in an applied electromagnetic field (FIG. 1b). Such a solution is restricted to explaining the most qualitative features of sinusoidal grating formation. We have established a general formulation of the problem which is capable in principle of handling gratings of any shape or depth.

In the H polarization, Maxwell's equations can be written as [22]

$$-\frac{\partial^{2} H}{\partial z^{2}} + \frac{\partial^{2} H}{\partial x^{2}} + \varepsilon(x) k_{o}^{2} H_{y} = \frac{\partial H}{\partial x} \frac{\partial}{\partial x} \left[\ln \varepsilon(x) \right], \qquad (7)$$

where H is the component of the magnetic field H in the direction perpendicular to the lattice vector of the grating (FIG. 1). For square-well gratings, H is separable,

$$H_{y}(x,y) = Z(z) X(x)$$
, (8)

and Eq. (7) can be written as two coupled first-order differential equations:

$$\frac{\partial z^2}{\partial z^2} = \Lambda^2 z \tag{9a}$$

$$\frac{\partial^2 X}{\partial x^2} - \frac{\partial}{\partial x} \left[\ln \varepsilon(x) \right] \frac{\partial X}{\partial x} + \left[\varepsilon(x) k_0^2 - \Lambda^2 \right] X = 0.$$
 (9b)

When the applied radiation is incident normal to the grating, it is found that Λ is the solution of the equation

10 not be in it to type in its.)
$$1 - \cos(\beta d/2) \cos(\alpha d/2) + 1/2(\epsilon \alpha/\beta + \beta/(\alpha \epsilon)) \sin(\beta d/2) \sin(\alpha d/2) = 0, \quad (10)$$

where d is the pariod of the grating, ϵ is the dielectric constant of the grating, $\alpha = (k^2 - \Lambda^2)^{1/2}$ and $\beta = (\epsilon k^2 - \Lambda^2)^{1/2}$.

The problem to be considered (FIG. 1c) is that of a multilayered grating.

In each layer the grating is periodic and the general solution to Eq. (9) for the n-th layer may be written as

$$\Psi_{\mathbf{n}}^{\mathbf{II}} = \sum_{k=1}^{10} \frac{\mathbf{n}_{k}(\mathbf{x}) \left[\mathbf{A}_{\mathbf{n}_{k}} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{e}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i}} \mathbf{e}^{\mathbf{i} \mathbf{a}^{\mathbf{i}} \mathbf{e}^{\mathbf{i}} \mathbf{e}^{\mathbf{i}} \mathbf{e}^{\mathbf{i}} \mathbf$$

where the coefficients A and B are to be determined by the boundary conditions between layers, and $\Lambda_{n\ell}$ is the ℓ -th solution to Eq. (10) for the n-th layer. By applying the boundary condition of continuity of H and $\frac{\partial}{\partial z}$ between layers, we can establish the recursion relations for A and B,

$$\vec{A}_{n} = \vec{E}_{n,n-1} \vec{A}_{n-1} + \vec{F}_{n,n-1} \vec{B}_{n-1}$$
 (12a)

$$\vec{B}_{n} = C_{n,n-1}\vec{A}_{n-1} + D_{n,n-1}\vec{B}_{n-1}, \qquad (12b)$$

where the matrices C, D, E, F are determined by the periodic structure of each layer. Furthermore, we can establish boundary conditions between the top two layers and the two lowest layers. In the topmost (infinite) layer, the eigensolution is written as [21]

$$\Psi^{I} = e^{-ik_0z} + \sum_{n=-\infty}^{\infty} R_n e^{ik_0[\gamma_n x + (1-\gamma_n^2)^{1/2}z]}.$$
 (13)

In the lowest (infinite layer) we have
$$\psi \mathbf{\Pi} = \sum_{n=-\infty}^{\infty} \mathbf{T}_n e^{i\mathbf{k}_0 [\gamma_n \mathbf{x} - (\varepsilon - \gamma_n^2)^{1/2} \mathbf{z}]} \tag{14}$$

where $\gamma_n = \frac{n\lambda}{d}$.

It is now possible to establish

$$--\vec{A}_{\hat{1}} = \vec{a}\vec{D} + \vec{b}\vec{R}^{-1} \qquad \text{for all } \vec{b} = \vec$$

$$\vec{B}_1 = b\vec{D} + a\vec{R}, \qquad (15b)$$

where D is a unit vector. In addition,

$$\vec{A}_{N} = L\vec{B}_{N}, \tag{16}$$

where N labels the penultimate (finite) layer, and the matrix L is determined by the parameters which characterize that layer. However, by applying the recursion relations of Eq. (12), it is possible to establish

$$\vec{A}_{N} = J_{1}\vec{D} + J_{2}\vec{R} \tag{17a}$$

$$\vec{B}_{N} = \kappa_{1} \vec{D} + \kappa_{2} \vec{R} \tag{17b}$$

Rearranging (16) and (17) gives

$$\vec{R} = (J_2 - : LK_2)^{-1} (LK_1 - J_1)\vec{D}, \qquad (18)$$

and by employing Eq. (12) once more we arrive finally at a matrix equation for the vectors \vec{A}_M and \vec{B}_M , in (15.2) and \vec{A}_M and \vec{B}_M

$$\begin{bmatrix}
\overrightarrow{A}_{M} \\
\overrightarrow{B}_{M}
\end{bmatrix} = \prod_{\substack{n=2\\ i}}^{M} \begin{bmatrix}
E_{n,n-1} & F_{n,n-1} \\
C_{n,n-1} & D_{n,n-1}
\end{bmatrix} \begin{bmatrix}
a & b \\
b & a
\end{bmatrix} \begin{bmatrix}
\overrightarrow{D} \\
\overrightarrow{R}
\end{bmatrix} \cdot (19)$$
After reduction to 75% in recognition will as

RESULTS AND DISCUSSION X 7 5 In them (13.1 x 19.3 cm.)

We have made a preliminary calculation for a square-well silver grating The field intensity $|\vec{E}(z,x)|^2$ is averaged over x at each of two values of z (z = 0 and z = -h), and the ratio R = $|\vec{E}(z=-h)|^2/|\vec{E}(z=0)|^2$ is plotted as a function of the well-depth (roughness parameter) h (FIG. 2). R is thus the ratio of intensity in the "peak" region of the grating to that in the "well" region. For very shallow gratings (h < 0.5 nm), or about two monolayers) it is seen that $|E(z=0)|^2 > |E(z=-h)|^2$, which supports the perturbation description of Brueck and Ehrlich. This would explain how a particular grating establishes itself out of the "noise" of microscopic roughness. It _also explains why the establishment of such a grating for low-power lasers requires the photodissociation of an organometallic compound and does not occur in the presence of metal vapor alone. However, at a certain depth (around 0.5 nm) $\frac{|E(z=-h)|^2}{|E(z=-h)|^2}$ begins to increase very rapidly. It seems here that the grating structure will only continue to reinforce itself by a mechanism other than photodeposition. Such a mechanism might be photoevaporation of the grating itself, whereby the trough regions are "excavated" by the field. Finally, we note that as $h \to \infty$, $|E(z = -h)|^2 \to 0$. This is consistent with an upper unit to the growth of the grating. Work is currently underway to determine whether these qualitative trends are consistent also with sinusoidal gratings.

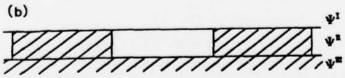
ACKNOWLEDGMENTS

This work was supported in part by the U.S. Army Research Office, the Office of Naval Research and the Air Force Office of Scientific Research (AFSC) under Grant No. AFOSR-82-0046. The United States Government is authorized to reproduce and distribute reprints for governmental purposes notwithstanding any copyright notation hereon. TFG acknowledges the Camille and Henry Dreyfus Foundation for a Teacher-Scholar Award (1975-84) and the John Simmon Guggenheim Memorial Foundation for a Fellowship (1983-84).

REFERENCES

- 1. M. Birnbaun, J. Appl. Phys. 36, 3688 (1965).
- 2. D. C. Emmony, R. P. Howson and L. J. Willis, Appl. Phys. Lett. 23, 598 (1977)
- 3. T. E. Zavecz and M. A. Saifi, Appl. Phys. Lett. 26, 165 (1975).
- 4. J. C. Koo and R. E. Slusher, Appl. Phys. Lett. 28, 614 (1976).
- 5. N. R. Isenor, Appl. Phys. Lett. <u>31</u>, 148 (1977).
- H. J. Leamy, G. A. Rozgonyi, T. T. Sheng and G. K. Celler, Appl. Phys. Lett. 32, 535 (1978).
- G. N. Maracas, G. L. Harris, C. A. Lee and R. A. McFarlane, Appl. Phys. Lett. 33, 453 (1978).
- 8. M. Oron and G. Sorensen, Appl. Phys. Lett. 35, 782 (1979).
- 9. A. K. Jain, V. N. Kulkarni, D. K. Sood and J. S. Uppal, J. Appl. Phys. 52, 4882 (1981)
- 10. P. A. Temple and M. J. Soileau, IEEE J. Quantum Electron. QE-17, 2067 (1981).
- 11. P. M. Fauchet and A. E. Siegman, Appl. Phys. Lett. 40, 1678 (1982).
- 12. S. R. J. Brueck and D. J. Ehrlich, Phys. Rev. Lett. 48, 1678 (1982).
- 13. J. F. Young, J. E. Sipe, J. S. Preston and H. M. van Driel, Appl. Phys. Lett. 41, 261 (1982).
- 14. H. M. van Driel, J. E. Sipe and J. F. Young, Phys. Rev. Lett. 49, 1955 (1982).
- J. E. Sipe, J. F. Young, J. S. Preston and H. M. van Driel, Phys. Rev. B 27, 1141 (1983).
- J. F. Young, J. S. Preston, H. M. van Driel and J. E. Sipe, Phys. Rev. B 27, 1155 (1983).
- J. F. Young, J. S. Preston, J. E. Sipe and H. M. van Driel, Phys. Rev. B 27, 1424 (1983).
- 18. For a comprehensive review, see A. A. Maradudin, in Surface Polaritons, ed. by V. M. Agranovich and D. L. Mills (North-Holland, Amsterdam, 1982) pp. 405-510.
- 19. Lord Rayleigh, Philos. Mag. 14, 60 (1907); Proc. R. Soc. London, Ser. A 79, 399 (1907).
- 20. A. Marin, F. Toigo and V. Celli, Phys. Rev. B <u>11</u>, 2777 (1974); S. S. Jha, J. R. Kirtley and J. C. Tsang, Phys. Rev. B <u>22</u>, **3973** (1982).
- 21. P. Sheng, R. S. Stepheman and P. N. Sanda, Phys. Rev. B <u>26</u>, 2907 (1982)
- M. Born and E. Wolf, <u>Principles of Optics</u> (Pergamon, New York, 1975), Chapt. 1.





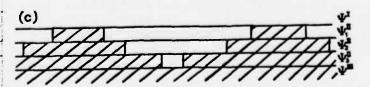


FIG. 1.

- (a) Sinusoidal grating. The hatched area represents the metal.
- (b) Square-well grating showing a separation into three layers, one of which is periodic in the x-direction, and two of which are uniform.
- (c) Generalization of the square-well grating in which there are three periodic layers.

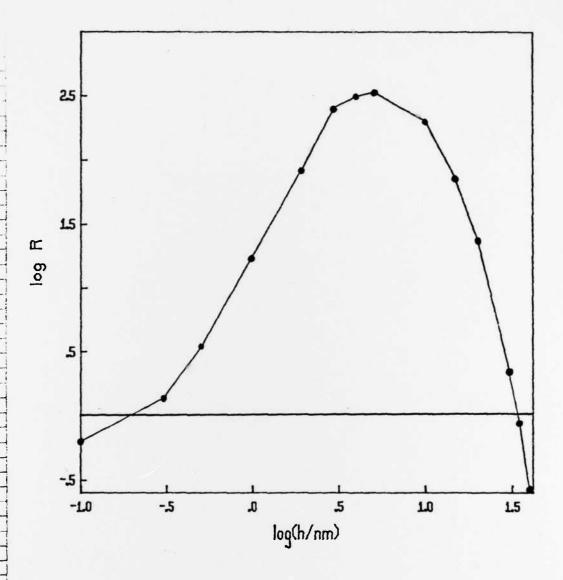


FIG. 2. Log-log plot of the ratio $R = |E(z=-h)|^2/|E(z=0)|^2$ against the grating depth h, for a silver grating with d=1050 nm and $\lambda=700$ nm. The horizontal line corresponds to R=1. Above the line the intensity is greatest in the region of the well (z=-h); below the line the intensity is greatest in the region of the peaks (z=0) of the grating. Clearly the latter condition pertains for very smooth gratings $(h \to 0)$ and also for very rough gratings $(h \to \infty)$

... of the from the lines of the place of the line.

TECHNICAL REPORT DISTRIBUTION LIST, GEN

	No. Copies		No. Copies
Office of Naval Research		Naval Ocean Systems Center	
Attn: Code 413		Attn: Mr. Joe McCartney	
800 North Quincy Street		San Diego, California 92152	1
Arlington, Virginia 22217	2		-
		Naval Weapons Center	
ONR Pasadena Detachment		Attn: Dr. A. B. Amster,	
Attn: Dr. R. J. Marcus		Chemistry Division	
1030 East Green Street		China Lake, California 93555	, 1
Pasadena, California 91106	1		
		Naval Civil Engineering Laboratory	
Commander, Naval Air Systems Command		Attn: Dr. R. W. Drisko	
Attn: Code 310C (H. Rosenwasser)		Port Hueneme, California 93401	1
Department of the Navy	•		
Washington, D.C. 20360	l	Dean William Tolles	
Different med 1 death T formant on him		Naval Postgraduate School	
Defense Technical Information Center		Monterey, California 93940	1
Building 5, Cameron Station	12	Scientific Advisor	
Alexandria, Virginia 22314	12		
Dr. Fred Saalfeld		Commandant of the Marine Corps (Code RD-1)	
Chemistry Division, Code 6100		Washington, D.C. 20380	1
Naval Research Laboratory		washington, b.o. 2000	•
Washington, D.C. 20375	1	Naval Ship Research and Development	
		Center	
U.S. Army Research Office		Attn: Dr. G. Bosmajian, Applied	
Attn: CRD-AA-IP		Chemistry Division	
P. O. Box 12211	•	Annapolis, Maryland 21401	1
Research Triangle Park, N.C. 27709	1	Mr. John Bootle	
Mr. Winner Cohone		Mr. John Boyle	
Mr. Vincent Schaper DTNSRDC Code 2803		Materials Branch Naval Ship Engineering Center	
Annapolis, Maryland 21402	1	Philadelphia, Pennsylvania 19112	1
Annapolis, Maryland 21402	•	rulladelpula, remusylvania 19112	•
Naval Ocean Systems Center		Mr. A. M. Anzalone	
Attn: Dr. S. Yamamoto		Administrative Librarian	
Marine Sciences Division		PLASTEC/ARRADCOM	
San Diego, California 91232	1	Bldg 3401	
		Dover, New Jersey 07801	1
Dr. David L. Nelson			
Chemistry Program			
Office of Naval Research			
800 North Quincy Street			
Arlington, Virginia 2217	1		

TECHNICAL REPORT DISTRIBUTION LIST, 056

Co	No.		No. Copies
		De II Vaha	
Dr. G. A. Somorjai		Dr. W. Kohn	
Department of Chemistry		Department of Physics University of California	
University of California	1	(San Diego)	
Berkeley, California 94720	1	La Jolla, California 92037	1
Dr. J. Murday			
Naval Research Laboratory		Dr. R. L. Park	
Surface Chemistry Division (6170)		Director, Center of Materials	
455 Overlook Avenue, S.W.		Research	
Washington, D.C. 20375	1	University of Maryland	- 1
		College Park, Maryland 20742	1
Dr. J. B. Hudson		Dog II M Dogs	
Materials Division		Dr. W. T. Peria	
Rensselaer Polytechnic Institute		Electrical Engineering Department	
Troy, New York 12181	1	University of Minnesota	1
na maria na Natao		Minneapolis, Minnesota 55455	1
Dr. Theodore E. Madey		Dr. Chia-wei Woo	
Surface Chemistry Section		Department of Physics	
Department of Commerce		Northwestern University	
National Bureau of Standards Washington, D.C. 20234	1	Evanston, Illinois 60201	1
washington, b.c. 20234	•	Brangton, 1111hold Gozof	•
Dr. J. M. White		Dr. Robert M. Hexter	
Department of Chemistry		Department of Chemistry	
University of Texas		University of Minnesota	
Austin, Texas 78712	1	Minneapolis, Minnesota 55455	1
Dr. Keith H. Johnson		Dr. R. P. Van Duyne	
Department of Metallurgy and		Chemistry Department	
Materials Science		Northwestern University	
Massachusetts Institute of Technology		Evanston, Illinois 60201	1
Cambridge, Massachusetts 02139	1		
		Dr. S. Sibener	
Dr. J. E. Demuth		Department of Chemistry	
IBM Corporation		James Franck Institute	
Thomas J. Watson Research Center		5640 Ellis Avenue	
P. O. Box 218		Chicago, Illinois 60637	1
Yorktown Heights, New York 10598	1		
		Dr. M. G. Lagally	
Dr. C. P. Flynn		Department of Metallurgical	
Department of Physics		and Mining Engineering	
University of Illinois		University of Wisconsin	,
Urbana, Illinois 61801	1	Madison, Wisconsin 53706	1

TECHNICAL REPORT DISTRIBUTION LIST, 056

	No. Copies		No. Copies
Dr. Robert Gomer		Dr. K. G. Spears	
Department of Chemistry		Chemistry Department	
James Franck Institute		Northwestern University	
5640 Ellis Avenue		Evanston, Illinois 60201	1
Chicago, Illinois 60637	1		
		Dr. R. W. Plummer	
Dr. R. G. Wallis		University of Pennsylvania	
Department of Physics		Department of Physics	
University of California, Irvine		Philadelphia, Pennsylvania 19104	1
Irvine, California 92664	1		
		Dr. E. Yeager	
Dr. D. Ramaker		Department of Chemistry	
Chemistry Department		Case Western Reserve University	
George Washington University		Cleveland, Ohio 41106	1
Washington, D.C. 20052	1		
		Professor D. Hercules	
Dr. P. Hansma		University of Pittsburgh	
Physics Department		Chemistry Department	
University of California,		Pittsburgh, Pennsylvania 15260	1
Santa Barbara		D. C. W. H.	
Santa Barbara, California 93106	1	Professor N. Winograd	
De I C Heerteen		The Pennsylvania State University	
Dr. J. C. Hemminger		Department of Chemistry	
Chemistry Department		University Park, Pennsylvania 16802	2 1
University of California, Irvine Irvine, California 92717	1	Drafagar W. E. Conne	
irvine, California 92/1/	1	Professor T. F. George	
Dr. Martin Fleischmann		The University of Rochester Chemistry Department	
Department of Chemistry		Rochester, New York 14627	1
Southampton University		Rochaster, New York 14027	_
Southampton SO9 5NH		Professor Dudley R. Herschbach	
Hampshire, England	1	Harvard College	
nampenite, ongrand	•	Office for Research Contracts	
Dr. G. Rubloff		1350 Massachusetts Avenue	
IBM		Cambridge, Massachusetts 02138	1
Thomas J. Watson Research Center		oumbridge, maddadmadeted viid	-
P. O. Box 218		Professor Horia Metiu	
Yorktown Heights, New York 10598	1	University of California,	
101100000	-	Santa Barbara	
Dr. J. A. Gardner		Chemistry Department	
Department of Physics		Santa Barbara, California 93106	1
Oregon State University			
Corvallis, Oregon 97331	1	Professor A. Steck1	
		Rensselaer Polytechnic Institute	
Dr. G. D. Stein		Department of Electrical and	
Mechanical Engineering Department		Systems Engineering	
Northwestern University		Integrated Circuits Laboratories	
Evanston, Illinois 60201	1	Troy, New York 12181	1

TECHNICAL REPORT DISTRIBUTION LIST, 056

	No. Copies	No. Copies
Dr. John T. Yates Department of Chemistry University of Pittsburgh		
Pittsburgh, Pennsylvania 15260	1	•
Professor G. H. Morrison Department of Chemistry Cornell University		
Ithaca, New York 14853	1	
Captain Lee Myers AFOSR/NC Bolling AFB Washington, D.C. 20332	1	
Dr. David Squire Army Research Office P. O. Box 12211		
Research Triangle Park, NC 27709	1	
Professor Ronald Hoffman Department of Chemistry Cornell University		
Ithaca, New York 14853	1	

